

Shrinkage Performance of Fly Ash Alkali-activated Cement Based Binder Mortars

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Abstract

Some authors reported that Alkali-activated Cement Based Binder (AACB) mortars can have much higher drying shrinkage than Portland cement based composites. Its worth remember that shrinkage performance is a very important property for reinforced concrete composites just because a high shrinkage performance is associated to cracking tendency that leads to future durability problems. Usually shrinkage is assessed under unrestrained conditions. However, the use of a restrained ellipse ring test is especially interesting for materials that will be used in restrained conditions like repair mortars. This paper provides results on restrained and unrestrained shrinkage performance of fly ash AACB mortars. The restrained shrinkage was assessed with an elliptical ring that provides a faster and more reliable assessment of the cracking potential than circular rings. The results show that the mixtures with lower sodium silicate content are associated to a lower unrestrained shrinkage. Mixtures with lower sodium silicate content showed reduced average crack width under restrained conditions. The results also show that the reduction of sodium silicate content delays the appearance of cracks and is also associated to lower crack width. The use of AACB mortars in restrained conditions requires the use of a reduced sodium silicate content or else the use of fibres to prevent crack appearance.

Keywords: waste reuse, AACB mortars, fly ash, capillary water absorption, shrinkage

1. Introduction

AACB are OPC alternative materials produced through the reaction of an aluminosilicate powder (precursor) with an alkaline activator, usually composed by hydroxide, silicate, carbonate or sulfate leading to the formation an amorphous aluminosilicate gel and secondary nano crystalline zeolite-like structures (van Deventer *et al.*, 2010, 2012; Pacheco-Torgal *et al.*, 2014; Provis, 2014).

These materials have a particular ability for the reuse of several types of waste (Payá *et al.*, 2014; Chindaprasirt and Cao, 2014; Bernal *et al.*, 2016). This is an important feature especially in the European context of a circular economy and zero waste target (COM, 2014) that may lead to an overall savings potential of several hundred billion euro per year for the European industry and can also create almost two hundred thousand direct jobs in the EU by 2030.

Some wastes like fly ash deserve an especial attention because they are generated in a very high amount and have a very low reuse rate. The reuse rate of fly ash in USA is around 50% meaning that million tons of fly ash are not reused annually (ACAA, 2016). Other recent works confirm the importance of further studies regarding the development of AACB based on fly ash precursors (Zhuang *et al.*, 2016).

Shrinkage performance is a very important property for reinforced concrete composites just because a high shrinkage performance is associated to cracking tendency that leads to future durability problems. Some authors (Collins and Sanjayan, 1999, 2000) reported that AACB composites can have much higher autogenous shrinkage than Portland cement based composites that could be explained by the fact that AACB composites showed a much higher mesopores content (82% versus 36%). Recent investigations confirmed this pore related explanation (Mobili *et al.*, 2016). However, AACB shrinkage is also dependent on the curing conditions being that curing with temperature is associated with a lower autogenous shrinkage (Wallah and Hardjito, 2014). Recent investigations showed that slag-fly ash AACB composites with a higher content of slag are associated with higher drying shrinkage (Gao *et al.*, 2016). This was also confirmed by Ye and Radlinska (2016) who suggested that shrinkage may be attributed to the structural incorporation of alkali cations in C-A-S-H, which reduces the stacking regularity of C-A-S-H layers and make the C-A-S-H easier to collapse and redistribute upon drying (Fig. 1).

Usually shrinkage performance is assessed through unrestrained shrinkage testing. Assessment of shrinkage using circular or ellipse restrained ring test is especially interest for materials that will be used in restrained conditions like repair mortars (Kristiawan, 2012; Khan, 2013; Beushausen and Chilwesa, 2013). Some

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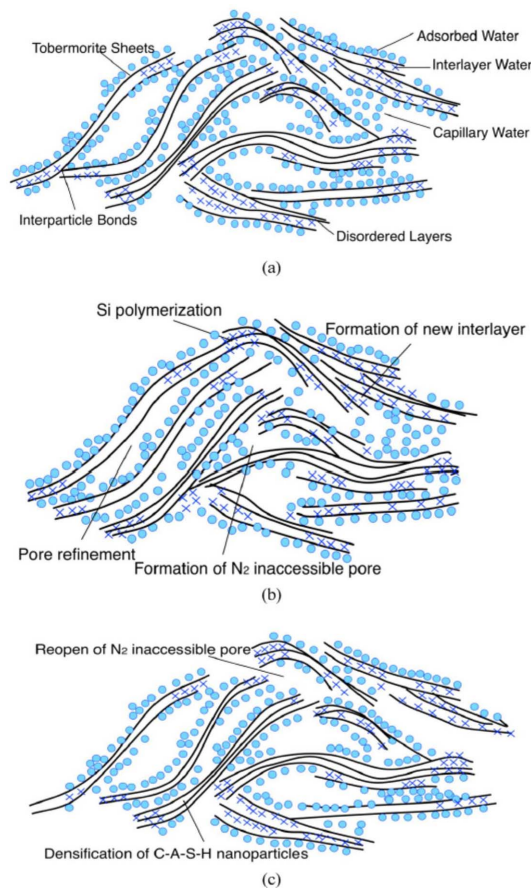


Fig. 1. Illustration of Nanostructure Changes of C-A-S-H During Drying: (a) Before Drying, CA-S-H Comprises of Primarily Amorphous Tobermorite-like Sheets, (b) After drying, CA-S-H Redistributes and Organizes; Rendering Pore Refinement, Silicate Polymerization, and New Interlayer Formation, (c) Dried at low RH (~11%RH), C-A-S-H Densifies and Loses Some Interlayer Water (Ye and Radlinska, 2016)

authors (Collins and Sanjayan, 2000) compared AACB and OPC concrete performance under restrained ring test reporting that the former cracked after only 1 week of curing, while the latter cracked only after 5 months. Ma and Ye (2015) recently published results comparing unrestrained and restrained shrinkage performance of fly ash AACB pastes stating that no cracks were detected. This paper provides results on restrained and unrestrained shrinkage performance of fly ash AACB mortars.

2. Experimental Program

2.1 Materials

The raw materials used for the preparation of the AACB mortars were fly ash, calcium hydroxide, kaolin, cement type I 42.5R, sand, sodium silicate, sodium hydroxide and superplasticizer. Solid sodium hydroxide obtained from commercially available product of ERCROS, S.A., Spain, was used to prepare a 12M NaOH solution. Distilled water was used to dissolve the sodium

Table 1. Chemical Composition of Major Oxides

Material	Oxides (wt.%)							
	SiO ₂	Al ₂ O ₃	Fe ₂ O ₃	CaO	MgO	Na ₂ O	K ₂ O	TiO ₂
Fly Ash	60.81	22.68	7.64	1.01	2.24	1.45	2.7	1.46

hydroxide flakes to avoid the effect of unknown contaminants in the mixing water. The NaOH mix was made one day prior to use in order to have a homogenous solution at the time of mortar preparation. The chemical composition of the sodium hydroxide was composed of 25%Na₂O and 75%H₂O. The sodium silicate was supplied by MARCANDE, Portugal. The chemical composition of the sodium silicate was of 13.5%Na₂O, 58.7% SiO₂ and 45.2%H₂O. The fly ash was obtained from the PEGO Thermal Power Plant in Portugal and it was classified as class F according to ASTM-C618 (2015) standard. The chemical composition of the fly ash selected for this study is presented in Table 1. The particle size distribution of the fly ash was varied from 2 μm to 59 μm. Fly ash was partially replaced by Portland cement type I class 42.5R from SECIL, Portugal and also by calcium hydroxide from LUSICAL H100. In term of chemical components, the cement contains 63.3% CaO, 21.4%SiO₂, 4.0%Fe₂O₃, 3.3%Al₂O₃, 2.4%MgO and other components. The calcium hydroxide used in this study contains more that 99% CaO. The sand was used as inert filler and was provided by MIBAL, Minas de Barqueiros, S.A. Portugal. Prior to its use it was sieved in order to remove dusts particles. The sieves used in this operation had a 4.75 mm and a 0.6 mm mesh size. The detailed grain size distribution of the aggregates presented in Fig. 3. The superplasticizer was supplied by BASF. It's a polycrylate from Acronal series, with a density of 1050 kg.m³.

2.2 Production of AACB Mortar Specimens

A sodium hydroxide solution (12M) was mixed with sodium silicate to dissolve the silica and alumina of the fly ash particles, resulting a homogenized gel lasting 1 min; next, all the solid materials were mixed by using standard mixer operating at the following speed I (65 rpm) for 3 min. After the addition of the alkaline activator the mixer was operated during 1min with speed I (65 rpm) and plus another 1 min with speed II (90 rpm). After the casting the unrestrained shrinkage specimens were cured at laboratory conditions with temperature about 25°C and 65% of relative humidity. After ≈4hr the specimens were demoulded and kept sealed with the plastic wrap and then left in the same curing conditions until the date of testing. Concerning restrained shrinkage, the AACB mortar was transferred to an elliptical mould. Then, this mould was placed on the vibration table for 2 min. The specimens were kept sealed by using plastic wrap and they were cured at laboratory conditions with temperature 20 ± 3°C and 70% ± 10% HR. After 24hr the specimen the outer elliptical ring was removed and then left in the same curing conditions. The specimens in the elliptical mould were monitored during 14days in order to detected crack formation on the surface of the specimens.

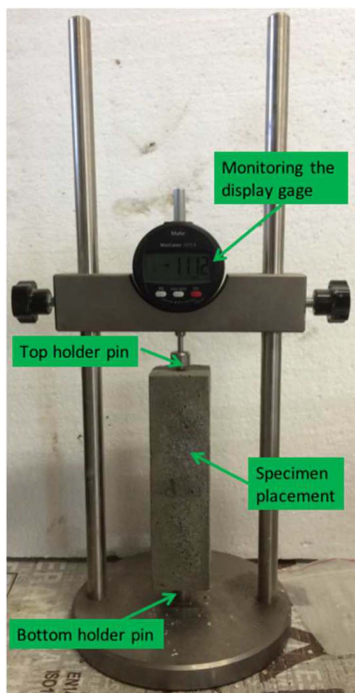


Fig. 2. Type of Used Digital Micrometer Apparatus for Measurement of Length Changes

2.3 Test Procedures

The following tests have been carried out: unrestrained shrinkage, restrained shrinkage, modulus of elasticity and capillary water absorption. A total of 6 mixtures were tested under both shrinkage tests. This included 4 representative mixtures made with fly ash (90%FA), calcium hydroxide (10%CH) and a sodium hydroxide activator (12M) with two activator/binder ratios (0.8 and 1.0), with two sodium silicate to sodium hydroxide mass ratios (2.0 and 2.5). The composition of the AACB mixtures can be found in Table 2.

2.3.1 Capillary Water Absorption

The capillary water absorption was performed according to EN 1015-18 standard. The specimens had the following dimensions 40 mm × 40 mm × 80 mm. The surface the specimens were coated with silicon in order to assure the water can be absorbed from one surface of the specimens that is soaked in water. The

quantification of absorbed water was performed by conducting weightings of the specimens through several segment intervals which started at 0 min, 5 min, 10 min, 20 min, 30 min, 60 min, 90 min, 120 min, 180 min, 240 min, 300 min, 360 min, 1440 min and 2880 min. However, it should be mentioned that, in order to assess the evolution of the water absorption towards weight stabilization, periodic measurements were performed for about 4days. The capillary absorption coefficient was determined according to the following (Eq. (2)):

$$C = \frac{M2 - M1}{\sqrt{tf - ti}} \quad (2)$$

In which, C is the capillary water absorption coefficient ($\text{kg}/(\text{m}^2 \cdot \text{min}^{0.5})$); $M2$ is the weight if the specimen in contact with water at the instant 10 min (kg/m^2); $M1$ is the weight if the specimen in contact with water at the instant of 90 min (kg/m^2); tf is the final time, instant of 90 min (min); and ti is the initial time, instant of 10 min (min).

2.3.2 Modulus of Elasticity

The modulus of elasticity test was based on cylindrical specimens (88 mm in diameter and 170 mm in length). 24hr prior to all testing, cylinders were capped using a stabilizer compound which provided a smooth surface for the platen to contact and reduced variability between compression specimens. The test was done according to the European standard EN 12390-13. Mortar cylinders from each mixture were tested under 100KN capacity testing frame. Computer software controlled the load rate and collected data for the applied force. Three Linear Variable Differential Transformer (LVDT) attached to a compressometer and wired to the data acquisition system and computer software as a standard strain gauges. Two rings were considered to be placed at the height of one-third and two-third of global height of the cylinder which the axis distances between the rings were considered fixed of 60 mm through all the tests. In fact, three LVDTs are rigidity attached upon top ring and are free to rotate on the horizontal surface of the bottom ring. Using loading cycled upon 40% of the ultimate stress, to obtain the equation of the straight line in the elastic zone for each specimen. The slope of this line in the static modulus of elasticity. Data for load and displacement was acquired automatically by the computer software, and the modulus of elasticity of each specimen was

Table 2. Mixture Composition of AACB Mortars

Sample name	Materials (kg/m^3)*								
	PC	MK	FA	CH	SA	SS	SH	SP	WT
90FA_10CH_12M_2.5S/H_0.8A/B	0	0	438.5	48,7	1461.5	278	111.7	0	0
90FA_10PC_12M_2.5S/H_0.8A/B	49	0	441.5	0	1471.3	280	112.5	0	0
90FA_10CH_12M_2.5S/H_1.0A/B	0	0	415.8	46.2	1385.9	330	132.4	0	0
90FA_10CH_12M_2.0S/H_0.8A/B	0	0	435	48.3	1449.5	257.7	128.8	0	0
90FA_10PC_12M_2.0S/H_0.8A/B	48.6	0	437.7	0	1459.2	259.4	129.7	0	0
90FA_10CH_12M_2.0S/H_1.0A/B	0	0	411.2	45.8	1372.3	304.9	152.5	0	0

*PC: stands for the Portland cement; MK: stands for the metakaolin; FA: stands for the fly ash; CH: stands for the calcium hydroxide; SA: stands for sand; SS: stands for sodium silicate; SH: stands for the sodium hydroxide; SP: stands for superplasticizer; WT: stands for the water.

calculated as the average of the three slopes of the stress-strain curves.

2.3.3 Unrestrained Shrinkage

The unrestrained shrinkage test used specimens with dimensions of 40 mm × 40 mm × 160 mm. The procedure of the length change assessment followed recommendation of ASTM C490-00 standard. The length changes measurements started at the second day from casting and continued up to 28 days. The specimens were cured in a control room condition at 20 ± 3°C and 70% ± 10% relative humidity. In order to calculate the length change, each specimen was placed into the apparatus model Mahr-MarCator 1075R (Fig. 2). Then the four sides of the specimens were monitored by rotating the specimen slowly in the measuring apparatus while the reading is taken and the mean value was considered. The length change of the specimens were reported when compared with the reference specimen with a digital micrometer in microstrain. A dimensionless parameter L has been introduced to evaluate the effect of controlled curing on the mass loss percentage (Eq. (1)):

$$L = \frac{\Delta w}{w_0} \quad (1)$$

Where Δw is the increment of weight loss and w₀ is the initial weight of the specimen.

2.3.4 Restrained Shrinkage

Restrained shrinkage tests were conducted using an elliptical ring test. Just because several authors (Zhou *et al.*, 2014; Dong *et al.*, 2014) have already demonstrated the advantages of using elliptical rings over circular ones in faster and more reliable assessment of cracking potential of cementitious composites. Also contrary to circular ring in which a crack may initiate anywhere, for an elliptical ring specimen, it tends to be at a determinable position, reducing the resources needed for detecting crack initiation/propagation in the ring test (Dong *et al.*, 2016). The dimensions of the elliptical ring test were the same of the one

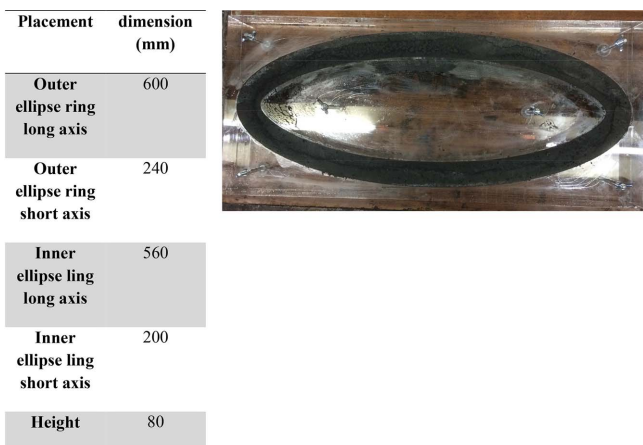


Fig. 3. Ellipse Ring Test Setup (including a table with dimensions of the ellipse sample)

used recently by Ma and Ye (2015). Its dimensions are shown in Fig. 3).

3. Results and Discussion

3.1 Capillary Water Absorption

Capillary water absorption coefficients as well as capillary water absorption with time are shown in Fig. 4 and Fig. 5. The capillary water absorption coefficients for AACB mortars range from ≈ 0.23 kg/m²min^{0.5} to 0.51 kg/m²min^{0.5}. This values are within the interval (0.1-0.9 kg/m²) already reported by other authors that studied alkali-activated binders (Pinto, 2004; Pacheco-Torgal *et al.*, 2008). A minor reduction on the capillary water absorption coefficient is noticed when the sodium silicate amount is reduced. This finding was already reported by other authors (Borges *et al.*, 2016). For the same mixture the reduction of the activator/binder from 1 to 0.8 is responsible for a reduction on the capillary water absorption coefficient of around 50%. The higher value was reported for the mixture with 90%FA plus 10% CH with the higher content of sodium silicate and a ratio activator/binder of 1.0. The replacement of 10% fly ash by Portland cement or by 10% calcium hydroxide show no relevant differences concerning the capillary water absorption.

3.2 Modulus of Elasticity

The results of the modulus of elasticity are shown in Table 3. The reduction of activator/binder ratio from 1.0 to 0.8 leads to high increase of the modulus of elasticity in the AACB mortars. As to the influence of OPC addition the results show that the

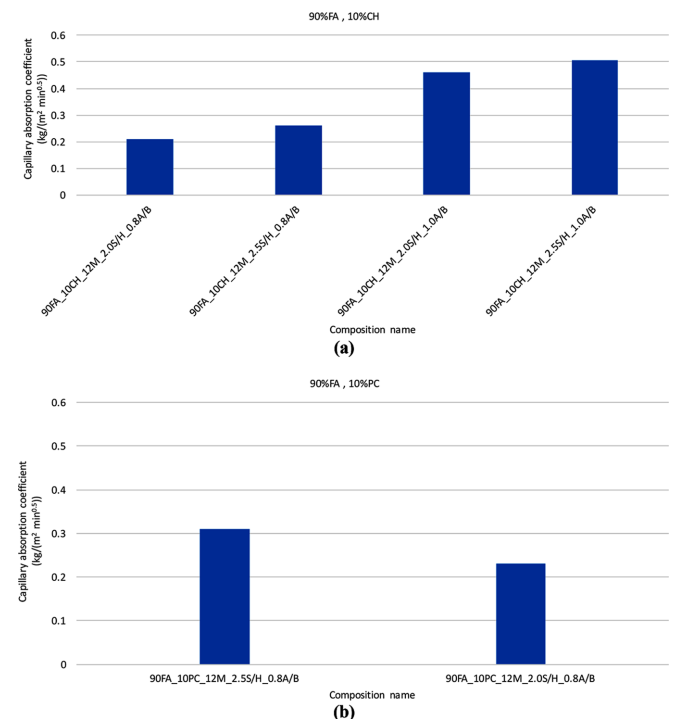


Fig. 4. Capillary Water Absorption Coefficient: (a) 90%FA, 10%CH, (b) 90%FA, 10%PC

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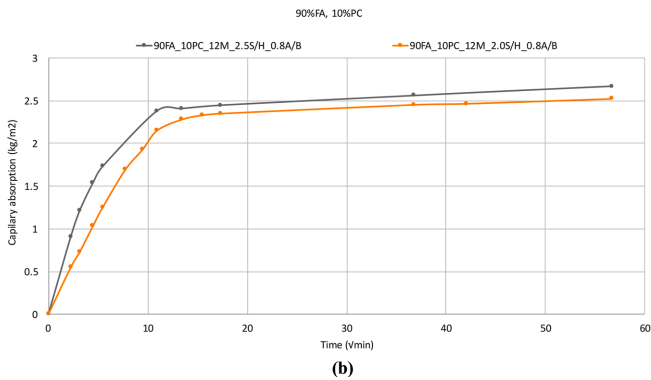
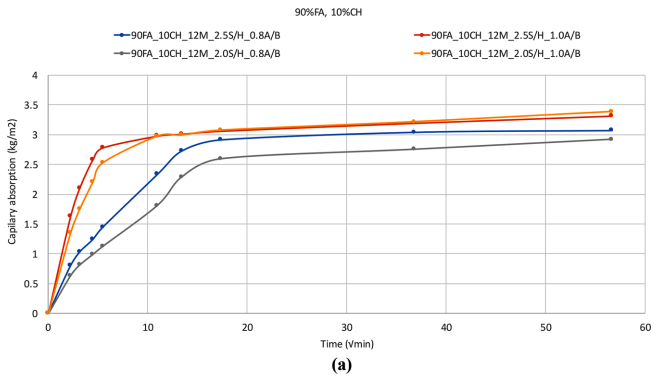


Fig. 5. Capillary Water Absorption with Time: (a) 90%FA, 10%CH, (b) 90%FA, 10%PC

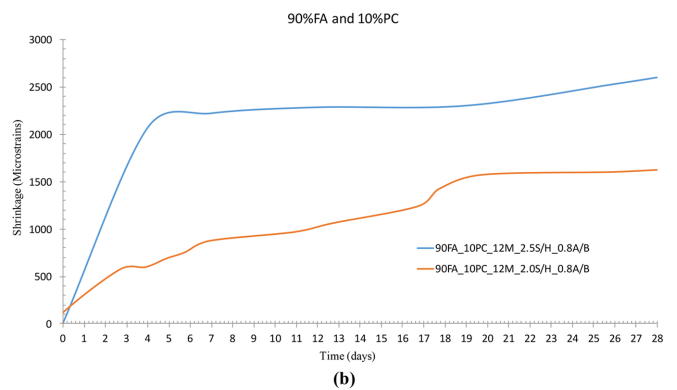
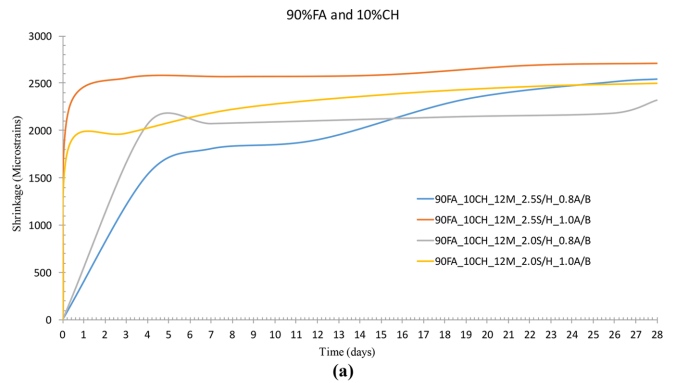


Fig. 6. Shrinkage of AACB Mortars with: (a) 90%FA and 10%CH, (b) 90%FA and 10%PC

Table 3. Modulus of Elasticity

Formulation	Modulus of elasticity (GPa)
90FA_10CH_12M_2.5S/H_0.8A/B	15.02
90FA_10PC_12M_2.5S/H_0.8A/B	12.28
90FA_10PC_12M_2.0S/H_0.8A/B	11.4
90FA_10CH_12M_2.0S/H_0.8A/B	8.8
90FA_10CH_12M_2.5S/H_1.0A/B	4.6
90FA_10CH_12M_2.0S/H_1.0A/B	3.8

related mixtures show a minor reduction of the modulus of elasticity when CH is used instead. This because such replacement is associated to a higher content of CSH (Pangdaeng *et al.*, 2014). AACB mixtures with a higher activator/binder ratio are associated with the lower modulus of elasticity (from 3.8 to 4.6 GPa). In this mixtures it is the increase of the sodium silicate/sodium hydroxide ratio that explains the increase in the modulus of elasticity. An increase of the modulus of elasticity with the increase of the Si was noticed by other authors that study alkali-activated based binders (Duxson *et al.*, 2005). These authors suggested that the greater lability of silicate species within the sodium silicate solution allows extensive gel reorganization and densification resulting in a microstructure comprising dense gel particles. The increase of the gel volume allows for a greater cross-section of gel to support mechanical loads explaining the increase of the modulus of elasticity.

3.3 Unrestrained Shrinkage

Unrestrained shrinkage results are shown in Fig. 6. The

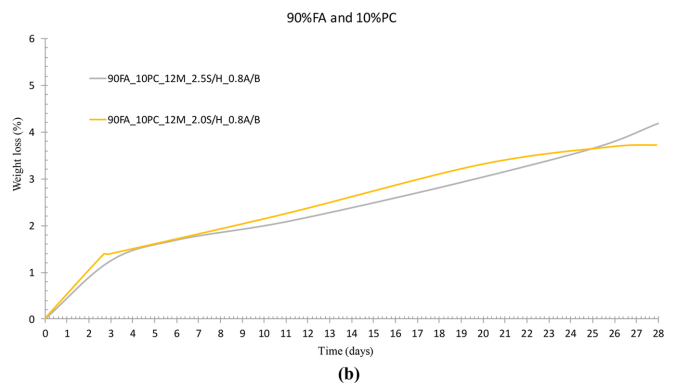
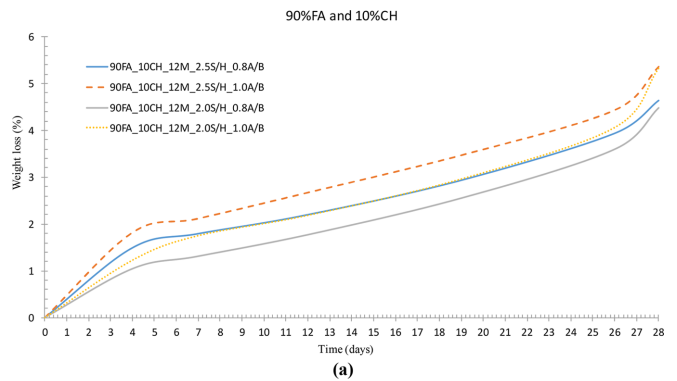


Fig. 7. Weight Loss for AACB Mortars Cured at Control Environment Temperature of $20^{\circ}\text{C}\pm 3^{\circ}\text{C}$ with R.H of $70\% \pm 10\%$: (a) 90%FA and 10%CH, (b) 90%FA and 10%PC

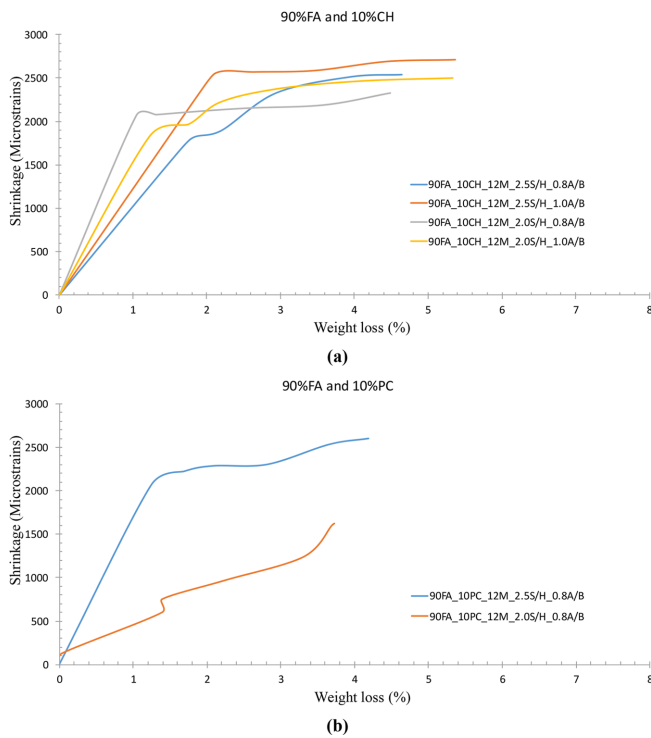


Fig. 8. Weight Loss as a Function of the Unrestrained Shrinkage for AACB Mortars: (a) 90%FA, 10%CH, (b) 90%FA, 10%PC

unrestrained shrinkage values exceed 1500 microstrain and can go as high as 2500 microstrain. The results of unrestrained shrinkage agree with the capillary water absorption just because the finer the capillary network the higher the capillary stress that generates shrinkage (Mobili *et al.*, 2016). A reduction of sodium silicate is associated with a lower shrinkage as already reported by other

authors (Gao *et al.*, 2016). Fig. 7 shows the weight loss of mixtures tested for unrestrained shrinkage. The percentage weight loss was monitored, bound to be less than 6% operating 28 days of curing for all the mixtures. The mixtures with higher SS/SH ratio show higher weight loss. Results show that mixtures with CH, SS/SH = 2.5 and A/B = 1.0 had shown the highest weight loss while, the ones with CH, SS/SH = 2.0 and A/B = 0.8 was the lowest, as expected. Fig. 8 shows restrained shrinkage versus the weight loss. The results show that two phases can be seen. A first one in which weight loss is proportional to unrestrained shrinkage that occurs for a weight loss below 1.5% (for mixtures with PC) and with CH and a lower sodium silicate content or below 2% for most mixtures with CH and a higher sodium silicate content. In this phase for the same weight loss of let's say 1% the lower unrestrained shrinkage is noticed for mixture 90FA_10CH_12M_2.5S/H_0.8A/B while the higher was reported for mixture 90FA_10CH_12M_2.0S/H_0.8A/B. A second phase is associated to a high water loss with minimal increase in unrestrained shrinkage. Meaning that the majority of water loss by evaporation has little influence on unrestrained shrinkage.

3.4 Restrained Shrinkage

3.4.1 Crack Initiation and Propagation

Figure 9 shows crack evolution for the mixture 90FA_10CH_12M_2.5S/H_0.8A/B. The first crack appeared after 32hr has propagated through the mortar ring wall (see Fig. 9(a), (b)), while at the 14 days of testing (time=336hr) the crack was totally separated. For mixture 90FA_10PC_12M_2.5S/H_0.8A/B (Fig. 10), the initial crack was noticed after 48 hours. Fig. 11 shows the crack positions of mixture 90FA_10CH_12M_2.5S/H_1.0A/B. As observed, crack originally initiated at mark I along the

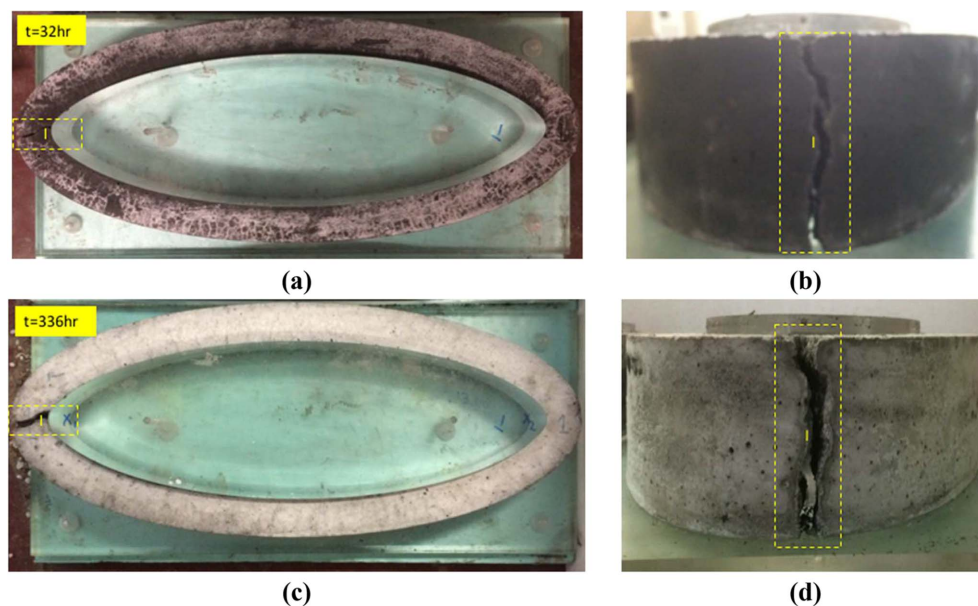


Fig. 9. Crack Positions in thin Elliptical Ring for the Mixture 90FA_10CH_12M_2.5S/H_0.8A/B: (a) Top View of the Crack Position at the Initial Time of Crack Appearance, (b) Lateral View of the Initial Crack Position, (c) Top View of the Crack Position at the Final Time of Test, (d) Lateral View of the Final Crack Position

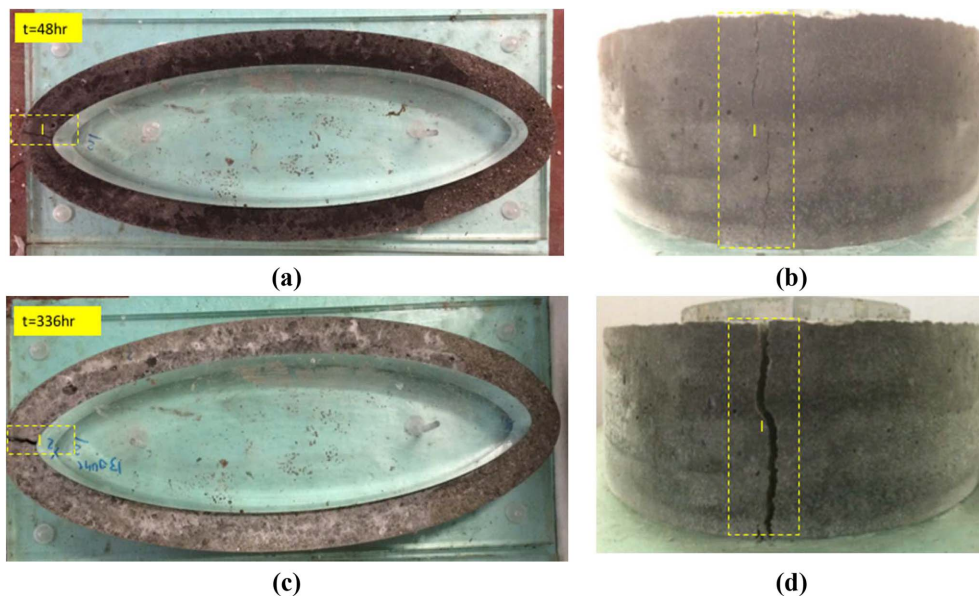


Fig. 10. Crack Positions in thin Elliptical Ring for Mixture 90FA_10PC_12M_2.5S/H_0.8A/B: (a) Top View of the Crack Position at the Initial Time of Crack Appearance, (b) Lateral View of the Initial Crack Position, (c) Top View of the Crack Position at the Final Time of Test, (b) Lateral View of the Final Crack Position

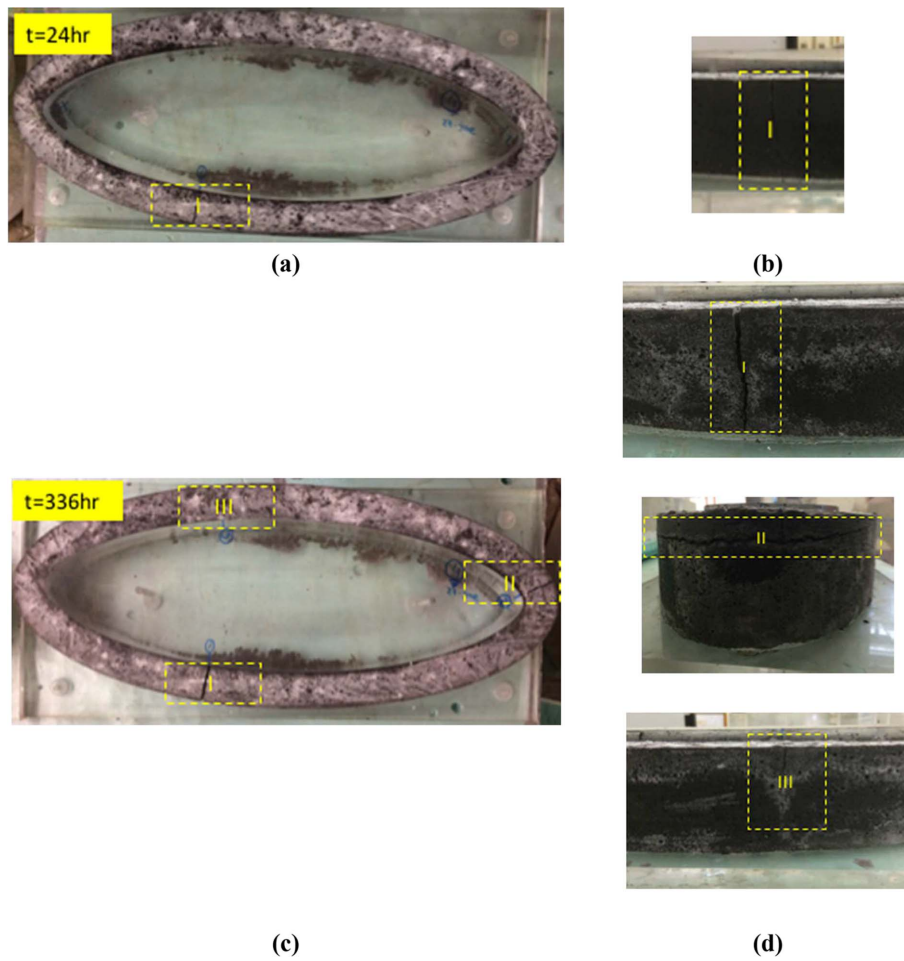


Fig. 11. Crack Positions in Thin Elliptical Ring for Mixture 90FA_10CH_12M_2.5S/H_1.0A/B: (a) Top View of the Crack Position at the Initial Time of Crack Appearance, (b) Lateral View of the Initial Crack Position, (c) Top View of the Crack Positions at the Final Time of Test, (b) Lateral View of the Final Crack Positions Composes of Three Different Crack Placement

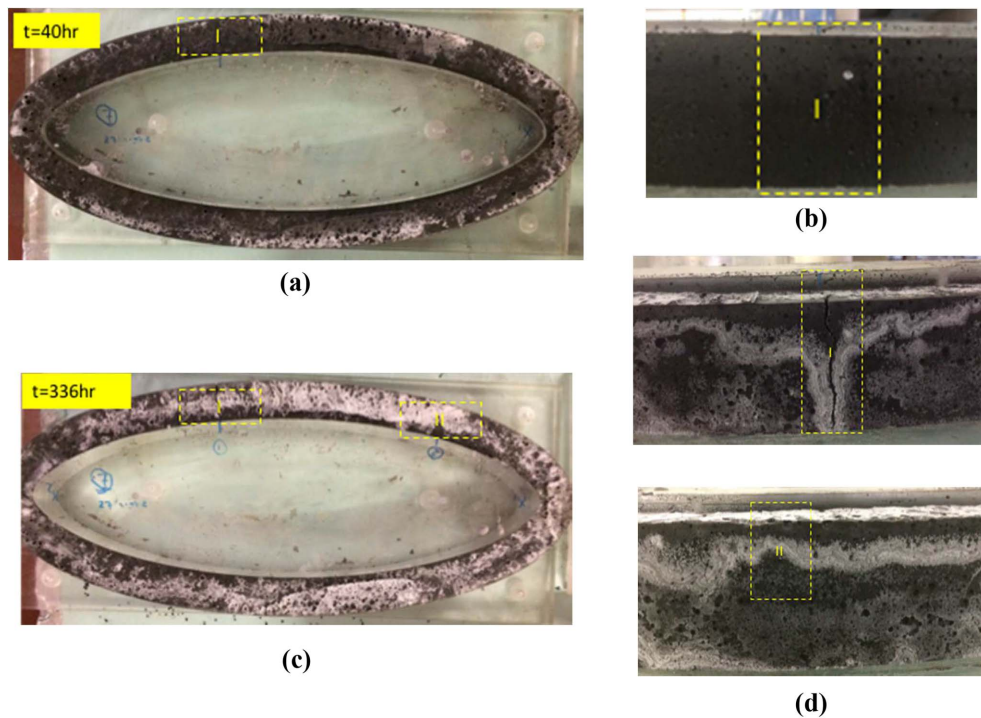


Fig. 12. Crack Positions in Thin Elliptical Ring for Mixture 90FA_10CH_12M_2.0S/H_0.8A/B: (a) Top View of the Crack Position at the Initial Time of Crack Appearance, (b) Lateral View of the Initial Crack Position, (c) Top View of the Crack Positions at the Final Time of Test, (d) Lateral View of the Final Crack Positions Composes of Two Different Positions

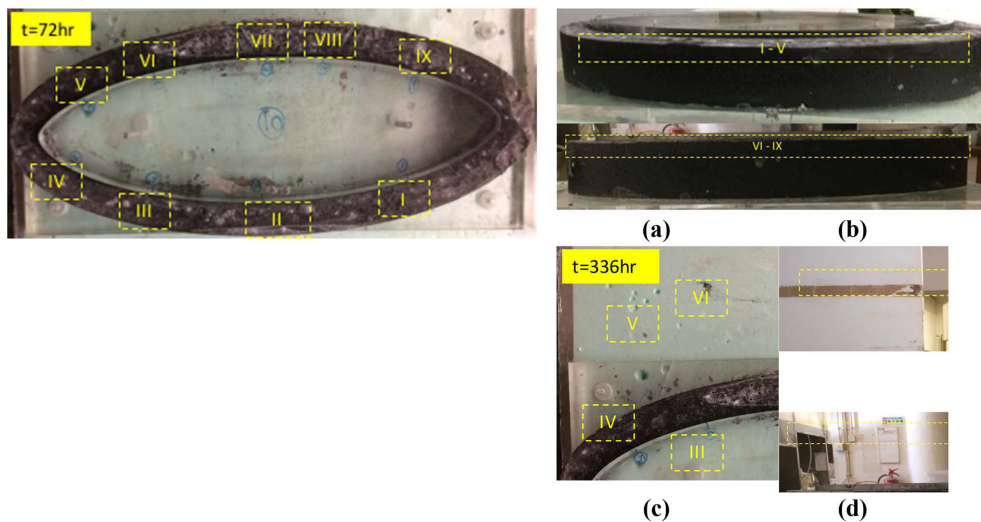


Fig. 13. Crack Positions in thin Elliptical Ring for Mixture 90FA_10CH_12M_2.0S/H_1.0A/B: (a) Top View of the Crack Position at the Initial Time of Crack Appearance, (b) Lateral View of the Initial Crack Position, (c) Top View of the Crack Position at the Final Time of Test, (d) Lateral View of the Final Crack Position Placed at Both Longitudinal Sides of the Elliptical Mortar Ring

height wall at the time of 24hr. Then, the cracks marked as II and III were appeared but crack III ceased to propagate due to reduced crack driving energy. However, as a result of the increased tensile stress redistribution, cracks were propagated at mark I and II. However, as a result of the increased stress in mark II, it is developed in the radial direction of the ring possibly it is due to the presence of moisture gradient (Hossain and Weiss, 2004). Fig. 12 indicates that for mixture 90FA_10CH_12M_2.0S/H_0.8A/

B the initial crack was observed at $t = 40\text{hr}$ at the region marked as I. Crack originally initiated at mark I along the height axis due to the stress concentration which caused the appearance of another crack at mark II. The experimental ellipse ring testing for mixture 90FA_10CH_12M_2.0S/H_1.0A/B (Fig. 13) show crack patterns different than earlier studied mortars. Initial cracks were observed after 72 hours (later than the other AACB mortars) at mark I up to IX. All the cracks shown in Fig. 13 were

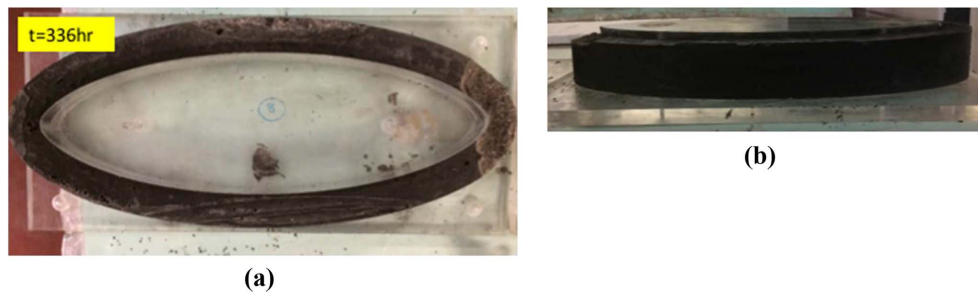


Fig. 14. No Visible Crack in thin Elliptical Ring for Mixture 90FA_10PC_12M_2.0S/H_0.8A/B at the Final Time of Test: (a) Top View, (b) Lateral View

Table 4. Visible Initial Cracking Time of the AACB Mortars

Sample name	Visible initial cracking time (hr)
90FA_10CH_12M_2.5S/H_0.8A/B	32
90FA_10PC_12M_2.5S/H_0.8A/B	48
90FA_10CH_12M_2.5S/H_1.0A/B	24
90FA_10CH_12M_2.0S/H_0.8A/B	40
90FA_10PC_12M_2.0S/H_0.8A/B	*
90FA_10CH_12M_2.0S/H_1.0A/B	72

*No visible crack has been monitored.

observed at top surfaces with almost similar sized initial crack. This may imply that the crack driving energy was lower for this mixture having required a higher tensile stress to lead a crack to propagate after it is initiated. As shown in Fig. 14, no visible cracks were observed on the surface of the mixture 90FA_10PC_12M_2.0S/H_0.8A/B after 336hr of testing. This result is related to a low modulus of elasticity as suggested by others (Ma and Ye, 2015; Mobili *et al.*, 2016).

3.4.2 Initial Cracking Time

The initial cracking time is given in Table 4. It is clear that the influence of the different binder on the initial cracking time. In fact, the crack appearance of AACB mortars was strongly affected by the SS/SH and A/B ratios. For the ratio of A/B = 0.8, the initial cracking time of mortars with PC apparently increased with increasing SS/SH ratio. Especially, no crack appeared for a period of 336hr on the mortar with PC with SS/SH = 2.0. The initial cracking time of mortars with CH enhanced with decreasing ratio from SS/SH = 2.5 to SS/SH = 2.0 regardless of A/B ratio. Mixtures with a high sodium silicate content, A/B = 0.8 and PC shown an initial cracking time enhanced by 16hr when compared with the mortar with CH. It seems that PC inclusion decreased the cracking sensitivity. It is further noticed that for mixture 90FA_10CH_12M_2.0S/H_1.0A/B, the initial crack was visible much later than that of CH mortar with the similar ratios of SS/SH or A/B.

3.4.3 Crack Width

The average value of the crack width is presented in Fig. 15. It can be seen that in all cases (except for mixture 90FA_10PC_

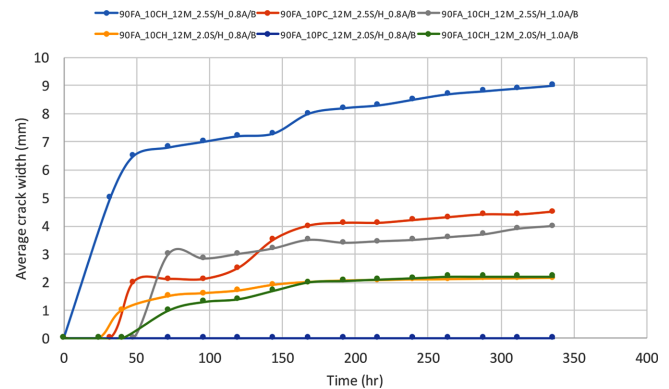


Fig. 15. Average Crack Width for Different Mortars During Tested Period of 14 days

12M_2.0S/H_0.8A/B), a gradual increase with age followed by a sudden and significant jump in the width of the cracks. As specimen dries, its shrinkage increases with age. However, enhancing shrinkage of the material is restrained by the central ellipse ring; resulting in an increase of the tensile stress which is imposed on the central ellipse ring. Simultaneously, tensile stress is developed in the mortar. When the tensile stress exceeds the tensile strength of mortar, crack initiates and the stress imposed by central ring is released. Comparing the crack width of mixture containing CH, a maximum crack width was noticed for a higher sodium silicate content and A/B = 0.8. The increase in the activator/binder ratio lead to a 55% reduction of the width crack at 14 days of testing. It is also noticed, for the ratio A/B = 0.8, that increasing sodium silicate content increase the average crack width by approximately 4.5 mm. The reduction of sodium silicate content delays the appearance of cracks and is also associated to lower crack width. Results also show that higher unrestrained shrinkage is associated with higher crack width.

3.5 Relationship between Unrestrained Shrinkage and Restrained Shrinkage

Figure 16 shows the relationship between average crack width (mm) and length change (microstrain) for different studied mortars. For most mixtures no clear correlation exists between the two parameters. However, for mixture 90FA_10PC_12M_2.5S/H_0.8A/B a clear correlation was detected. This mixture has not only a high sodium silicate/sodium hydroxide ratio but

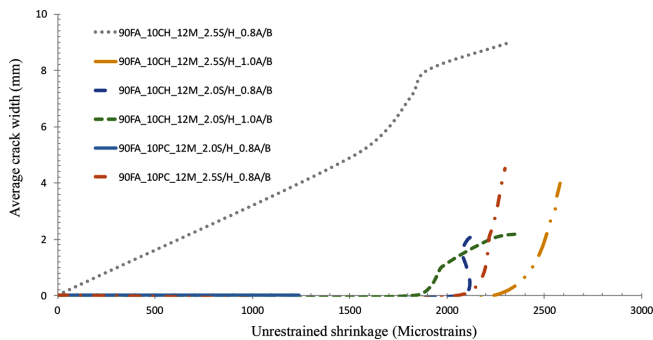


Fig. 16. Relationship between Average Crack width and Unrestrained Shrinkage

also a reduced activator/binder ratio the variables that are prone to the worst performance concerning the use of AACB mortars for restrained conditions. This means that for such conditions the use of fibres maybe needed to prevent crack appearance.

4. Conclusions

Some authors reported that alkali-activated cement based binder (AACB) mortars can have much higher drying shrinkage than Portland cement based composites. Its worth remember that shrinkage performance is a very important property for reinforced concrete composites just because a high shrinkage performance is associated to cracking tendency that leads to future durability problems. This paper provides results on restrained and unrestrained shrinkage performance of fly ash AACB mortars. The restrained shrinkage was assessed with the use of an elliptical ring. This device provides a faster and more reliable assessment of cracking potential than circular rings. Based on results of the testing program the following conclusions can be made. Mixtures with lower sodium silicate content are associated to lower unrestrained shrinkage. Weight loss results show that the majority of water loss by evaporation has little influence on unrestrained shrinkage. Elliptical ring tests revealed, that most of the crack initiated at the vertices on the axis and propagated from the top surface to the outer of the wall. Mixtures with CH and lower SS/SH ratio exhibited an older visible initial cracking time. While, in the specimens with PC at lower SS/SH ratio did not cause any visible crack during 14 days of testing. Results show that higher unrestrained shrinkage is associated with higher crack width. The reduction of the activator/binder ratio is associated to higher modulus of elasticity and higher restrained shrinkage. The reduction of sodium silicate content delays the appearance of cracks and is also associated to lower crack width. The use of AACB mortars in restrained conditions requires the use of a reduced sodium silicate content or else the use of fibres to prevent crack appearance.

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